#### **TL-Spectrometry and Applications in Biomedical Research and Diagnostics**

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### Molecular energy diagram and related excitation/deexcitation processes

Nonradiative modes of relaxation (C, B)

A, R, F: absorbance, resonant and R A nonresonant fluorescence I, P: intersystem crossing, phoshorescence **C**, **B**: internal conversion. 3 vibrational relaxation



### **Basics of thermal lens effect**

- During non-radiative relaxation of excited species temperature in the sample increases (10<sup>-4</sup> 10<sup>-3</sup> K)
- a temperature gradient is generated with maximum temperature at the axis of the excitation beam
- the resulting refractive index gradient acts as a lens (mostly: dn/dT < 0, diverging lens)
- laser beam is defocused (single beam or pump/probe configuration)
- beam radius and its intensity at the beam axis changes
- relative change in the beam intensity is proportional to the absorbance of the sample and to the power of the excitation beam.









#### Nonsteady thermal diffusion

$$\frac{\partial T(r,t)}{\partial t} = D\nabla^2 T(r,t) - v_x \frac{\partial T(r,t)}{\partial x} + \frac{1}{\rho C_p}Q(r,t)$$

- T(r,t)..... temperature
- D......thermal diffusivity
- ρ.... density
- c<sub>p</sub>.....heat capacity
- $\dot{\mathbf{Q}}(\mathbf{r},\mathbf{t})$ .....source term ("heat")
- $v_x$ ......velocity of the medium in x direction
- By solving nonsteady the thermal diffusion equation, changes in refractive index and related TLS signal can be calculated for different beam geometries and excitation regimes (pulsed, cw)



#### Pulsed and cw excitation with a Gaussian beam

• Pulsed:  

$$Q(r,t) = \frac{2\alpha E_0}{\pi a^2 t_0} \exp\left[-2(x^2 + y^2)/a^2\right]$$

• cw:  

$$Q(r,t) = \frac{2\alpha P_{av}}{\pi a^2} \left\{ \exp\left[-2(x^2 + y^2)/a^2\right] \right\} \times (1 + \cos \omega t)$$

 $E_0 \dots$  pulse energy a .... pump laser beam radius $t_0 \dots$  pulse width $P_{av} \dots$  cw laser average power $\alpha \dots$  absorbance (cm<sup>-1</sup>) $\omega \dots$  modulation frequency



#### **Thermal lens signal**

$$s(t) = \frac{w_2^2(t) - w_2^2(0)}{w_2^2(0)}$$

- $w_2(0)$ ....radius of an unperturbed probe beam at the detector site
- $w_2(t)$ ....time dependent radius of a probe beam perturbed by the thermal lens
- $w_0$ .....radius of the probe beam at its waist

$$w_{2}^{2}(t) = w_{0}^{2} \left[ \left( 1 - \frac{z_{2}}{f(t)} \right)^{2} + \frac{1}{z_{0}^{2}} \left( z_{1} + z_{2} - \frac{z_{1}z_{2}}{f(t)} \right)^{2} \right]$$



#### Simplifications for usual far field experimental configuration

• 
$$z_2 >> z_1, z_2 >> z_0 = \pi w_0^2 / \lambda$$

- $f(t) >> z_1, f(t) >> z_0$
- @ t=0, f(0)=∞
  - $-\lambda$  .....probe beam wavelength
  - $-z_0$ ....confocal distance

$$\mathbf{s}(\mathbf{t}) = -\frac{2\mathbf{z}_1}{\mathbf{f}(\mathbf{t})}$$



# Refractive index change and focal distance of thermal lens

$$n(x, y, t) = n_0 + \left(\frac{\partial n}{\partial T}\right)_{T_A} \times T(x, y, t)$$
  
•  $n_0$ ..... unperturbed refractive index at ambient  
temperature  $T_A$ 

collinear:

transversal:

$$\frac{1}{f} = -\frac{\partial n}{\partial T} \int_{-\infty}^{\infty} \left( \frac{\partial^2 T}{\partial x^2} \right) dy$$

- $\frac{1}{f} = -\frac{\partial n}{\partial T} \left\{ \left( \frac{\partial^2 T}{\partial r^2} \right) \frac{1}{f} \right\}$ • f....thermal lens focal length
- *l*....interaction length



# TLS signal for collinear configuration

• Pulsed: 
$$(t_0 \rightarrow 0)$$

$$s(t) = -\frac{4AE_0 z_1 (\partial n / \partial T)}{\pi k a^2 t_c} \frac{1}{(1 + 2t / t_c)^2}$$
  
• cw:  

$$s(t) = -\frac{2AP z_1 (\partial n / \partial T)}{\pi k a^2} \frac{1}{(1 + t_c / 2t)}$$
  

$$- t_c \dots \text{ time constant} = a^2 \rho c_p / 4k = 4a^2 D$$
  

$$- k \dots \text{ thermal conductivity of the sample}$$



### TLS signal for transversal configuration

• Pulsed:  $(t_0 \rightarrow 0)$ 

$$s(t) = -\frac{2\alpha E_0 z_1 (\partial n / \partial T)}{\sqrt{2\pi} kat_c} \frac{1}{(1+2t/t_c)^{3/2}}$$
  
• cw:  
$$s(t) = -\frac{2\alpha P z_1 (\partial n / \partial T)}{\sqrt{2\pi} ka} \frac{1}{(1+t_c / 2t)^{1/2}}$$



#### **TLS signal form**





### TLS signal in a single beam experiment

- $P/a^2$  changes with increasing  $z_1$
- the signal maximum is found at  $z_1 = z_0$ (parabolic model)  $z_0 = \pi a_0^2 / \lambda$

$$s(t) = -\frac{AP(\partial n / \partial T)}{\lambda k} \frac{1}{(1 + t_c / 2t)}$$
  
or at  $z_1 = z_0 \sqrt{3}$  (aberrant model)  
$$s(t) = -\frac{AP(\partial n / \partial T)}{\lambda k} \tan^{-1} \left[ \frac{1}{(1 + t_c / t)\sqrt{3}} \right]$$



### **E - Enhancement factor in TLS**

$$\frac{\Delta I}{I} = \frac{2.303P(-dn/dT)A}{\lambda k} arctg[1/\sqrt{3}] = 2.303EA$$

Solvent	$-dn/dT (10^4 \text{ K}^{-1})$	$k (W m^{-1} K^{-1})$	$E (10^{-3} \text{ W}^{-1})$
H <sub>2</sub> O	0.91	0.607	0.12
CCl <sub>4</sub>	5.9	0.103	4.74
acetone	5.42	0.190	2.36

 $E = (-dn/dT) / (1.91 \ \lambda k)$  is calculated for  $\lambda = 632.8 \text{ nm}$ 

#### Thermo-optical properties of solvents for TLS measurements

Solvent	Thermal conductivity, k mWcm <sup>-1</sup> K <sup>-1</sup>	10 <sup>4</sup> (dn/dT) K <sup>-1</sup>	$-\frac{10^4(dn/dT)}{k}$ cm mW <sup>-1</sup>
CO <sub>2 (SC)</sub>	0.7	-100	143
CCl <sub>4</sub>	1.03	-5.9	5.73
Benzene	1.24	-6.4	5.16
C <sub>8</sub> MImTf <sub>2</sub> N	n.d.	n.d.	4.55
cyclohexane	1.24	-5.4	4.35
BMImBF <sub>4</sub>	1.78	-7.54	4.24
n-heptane	1.26	-5.0	3.97
BMImTf <sub>2</sub> N	1.06	-4.0	3.78
dioxane	1.39	-4.6	3.31
EMImTf <sub>2</sub> N	n.d.	n.d.	2.37
methanol	2.20	-4.7	2.14
water	6.11	-0.8	0.13

Calc. values (except CO2) taken from Chieu D. Tran and T. A. Van Fleet, Anal. Chem. <u>60</u>, (1988) 2478



#### TLS - advantages

- High sensitivity
  - signal proportional to excitation laser power
  - absorbances as low as 10<sup>-7</sup> can be measured
- Enables On-line detection
  - fast response of TLS signal (on µs to ms time scale)
- Capability of measuring small samples
  - sub-pL volumes can be probed
  - detection in microfluidic systems



### **TLS – drawbacks and solutions**

- Sensitivity still needs improvement
  - Higher laser power? (photo-labile compounds)
  - Modify solvents
- Limited availability of laser sources
  - Coloring reactions, indirect detection
- Poor selectivity
  - Single wavelength measurements
  - Coupling to separation techniques (HPLC, IC, CE)
- Photodegradation
  - Measure in flowing systems

#### Dual beam TLS spectrometer for detection in FIA, HPLC and bioassays





#### **Adjustable beam size/position TLM**



#### **Temperature dependent TLS signal in**

water



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#### The effect of photosensitivity on TLS signal (case of Cr-DPC)





#### **HPLC-TLS degtermination of**



# The role of BTL in the transport of antioxidants across the cellular wall



### Improvement of selectivity by separation techniques (HPLC, IC)



LOD: 90 pM LOQ: 250 pM

Martelanc M., Žiberna L., Passamonti S., Franko M.: Anal. Chim. Acta 809, 2014, 174–182.

### Free bilirubin in blood serum samples



# Simultaneous determination of bilirubin and biliverdin



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# First detection and modulation of bilirubin in vascular endothelial cels



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## Advantages of TLS: extremely high sensitivity, small sample capability

(100 – 1000 times lower LOD than SF)





### **Bioanalytical FIA system**





#### **FIA-ELISA-TLS Detection of**

#### **Food Allergens**



- TLS signal proportional to the amount of allergen retained on the immunocolumn
- Analysis time < 8 min



incubation



cking

Step 4-2° Ab incubation

Ag

Step 3-Ag

incubation

Step 5-Enzyme incubation

Ag

Step 6-Product formation



#### **Determination of BLG and OVA by FIA-ELISA-TLS**



LOD for beta-lactoglobulin (BLG) =2.3 pg/ 100  $\mu$ L LOD for ovalbumin (OVA) =1 ng/ 100  $\mu$ L

(190 pg by ELISA – Bethyl) (1 µg by ELISA – Abcam)

## Determination of NGAL a biomarker of acute kidney injury







#### Thermal Lens Spo

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## PHOTOTHERMAL SPECTROSCOPY METHODS FOR CHEMICAL ANALYSIS

#### Stephen E. Bialkowski

Volume 134 in Chemical Analysis: A Series of Monographs on Analytical Chemistry and Its Applications

#### etric Detection in Flow Injection eparation Techniques

den Franko y for Environmental Research, P.O.Box 301, 'a Gorica, Slovenia

w injection analysis, liquid chromatography,





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